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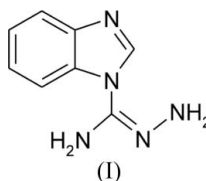
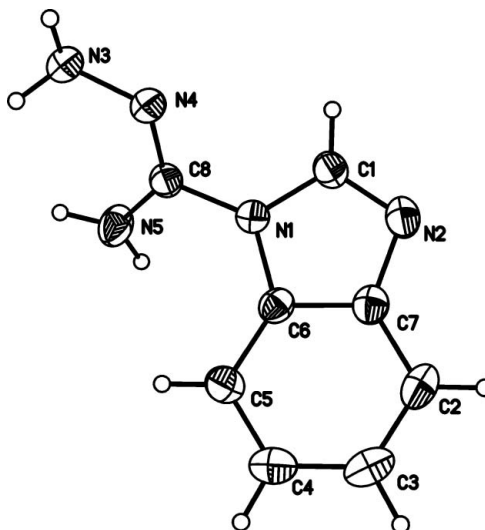
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## Key indicators

Single-crystal X-ray study  
 $T = 293$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.046  
 $wR$  factor = 0.143  
Data-to-parameter ratio = 15.2For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.1*H*-Benzimidazole-1-carbohydrazonamideThe title compound,  $\text{C}_8\text{H}_9\text{N}_5$ , was prepared by the reaction of 1-cyanobenzimidazole with a twofold excess of hydrazine. The crystal packing is stabilized by weak intermolecular  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds and  $\pi-\pi$  interactions.Received 7 June 2006  
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## Comment

*N*-Substituted benzimidazole (BzIm) derivatives are widely used in preparative organic syntheses (Staab, 1962). Usually, the BzIm derivatives are not stable in solution (Staab, 1962). Recent studies of *N*-benzimidazole carbohydrazonamides (Pan'kov *et al.*, 2001*a,b*) have shown their stability under normal atmospheric pressure in solution and in the solid state for a period of one month and more. The title compound, (I), was prepared by the reaction of 1-cyanobenzimidazole with hydrazine with a twofold excess of hydrazine. Here we present its crystal structure.The bond lengths and angles in (I) (Fig. 1) are normal (Allen *et al.*, 1987). In the crystal structure, the N atoms of the carbohydrazonamide group form four weak intermolecular  $\text{N}-\text{H}\cdots\text{N}$  hydrogen bonds with three neighboring molecules

**Figure 1**  
A view of (I) with the atom numbering. Displacement ellipsoids are drawn at the 50% probability level.

(Table 1), leading to the formation of a three-dimensional hydrogen-bonded network. The short intermolecular contact  $C2 \cdots C6^{iv}$  of 3.445 (8) Å indicates the presence of stacking interactions [symmetry code: (iv)  $1 - x, \frac{3}{2} + y, \frac{5}{2} - z$ ], which contribute to the stabilization of the crystal structure (Fig. 2) along with the  $N-H \cdots N$  hydrogen bonds.

### Experimental

1*H*-Benzimidazole-1-carbohydrazonamide was prepared according to the known procedure (Pan'kov *et al.*, 2001a). A portion of hydrazine hydrate (0.2 ml, 4.1 mmol) was added to a solution of 1-cyanobenzimidazole (0.3 g, 2.1 mmol) in 3 ml of 2-propanol. The reaction mixture was stirred and the temperature increased to 313–323 K. After 15 min it was cooled to room temperature and filtered. The product was washed with 2-propanol and recrystallized from ethanol. Yield 91%, m.p. 436–437 K.

#### Crystal data

$C_8H_9N_5$	$Z = 4$
$M_r = 175.20$	$D_x = 1.418 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 7.389 (2) \text{ \AA}$	$\mu = 0.10 \text{ mm}^{-1}$
$b = 8.152 (2) \text{ \AA}$	$T = 293 (2) \text{ K}$
$c = 13.660 (3) \text{ \AA}$	Prism, colorless
$\beta = 94.25 (3)^\circ$	$0.60 \times 0.20 \times 0.20 \text{ mm}$
$V = 820.5 (3) \text{ \AA}^3$	

#### Data collection

Siemens P3/PC diffractometer	$R_{int} = 0.025$
$\omega/2\theta$ scans	$\theta_{max} = 27.1^\circ$
Absorption correction: none	2 standard reflections
1929 measured reflections	every 98 reflections
1795 independent reflections	intensity decay: 2%
1448 reflections with $I > 2\sigma(I)$	

#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0798P)^2 + 0.29P]$
$R[F^2 > 2\sigma(F^2)] = 0.046$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.143$	$(\Delta/\sigma)_{max} = 0.011$
$S = 1.10$	$\Delta\rho_{max} = 0.35 \text{ e \AA}^{-3}$
1795 reflections	$\Delta\rho_{min} = -0.21 \text{ e \AA}^{-3}$
118 parameters	
H-atom parameters constrained	

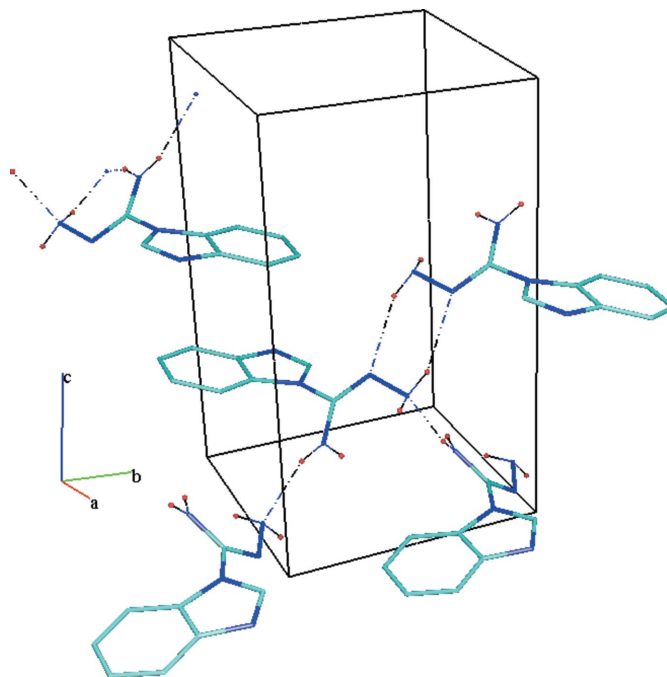
**Table 1**

Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$N3-H6 \cdots N2^i$	0.90	2.30	3.181 (2)	165
$N3-H7 \cdots N4^{ii}$	0.90	2.37	3.224 (2)	158
$N5-H9 \cdots N3^{iii}$	0.90	2.21	3.069 (2)	160
$N5-H8 \cdots N2^i$	0.90	2.23	3.083 (2)	158

Symmetry codes: (i)  $x + 1, y, z$ ; (ii)  $-x + 2, -y, -z + 2$ ; (iii)  $-x + 2, y + \frac{1}{2}, -z + \frac{3}{2}$ .

All H atoms were located in a difference map and refined as riding with  $C-H = 0.93 \text{ \AA}$ ,  $N-H = 0.90 \text{ \AA}$ , and  $U_{iso}(H) = 1.2U_{eq}(\text{parent atom})$ .



**Figure 2**

The crystal packing in (I). Hydrogen bonds are shown as dashed lines. The H atoms not participating in hydrogen bonds have been omitted for clarity.

Data collection: *P3/PC* (Siemens, 1989); cell refinement: *P3/PC*; data reduction: *XDISK* (Siemens, 1989); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL* and *TOPOS* (Blatov *et al.*, 1999); software used to prepare material for publication: *SHELXTL*.

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